Drug Chemistry Section
Drug Chemistry Procedure Manual
Effective Date: August 3, 1998

Modification of A-9
Prepared By: I.L. Allcox
Approved By: I.L. Allcox

Supersedes: September 1, 1996

## Name of Procedure:

Preliminary Tests Mecke's Reagent

## Suggested Uses:

The Mecke Color test consists of a solution of selenious acid and concentrated sulfuric acid. Aromatic compounds that typically undergo oxidation/reduction/substitution reactions will react with this reagent to produce colored intermediates. Refer to pages 631-649, "Spot Tests: A Color Chart Reference for Forensic Chemists", (see **Literature References**) for color formations of various drugs.

# **Apparatus Needed to Perform Procedure Including Preparation of Reagent:**

Fume hood
Gloves
Eye protection
Laboratory coat
Pipet with bulb
Graduated cylinder
50ml beaker
Glass stirring rod
Sulfuric acid (concentrated)
Selenious acid
Funnel
Reagent bottle
Porcelain spot plate
Spatula

## Formula for Preparing Reagent:

- 1. Weigh out 0.25 gram of selenious acid.
- 2. Dissolve in 25 milliliters of concentrated sulfuric acid.

# Formula for Preparing Reagent (Continued):

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3. Pour solution into a reagent bottle.

4. Properly label reagent bottle.

#### **Quality Control:**

A quality control check of this reagent will be performed using a known standard of heroin and following the application procedure listed below.

### **Expiration Date of Chemical Reagent:**

The Mecke's reagent should be prepared every 30 days.

### **Application of Procedure on Evidence:**

- 1. Place 1-2 drops of the reagent into a clean well on a spot plate.
- 2. With a spatula, add approximately 0.1 milligram of the unknown powder/tablet to the reagent in the spot plate.
- 3. Observe 1-2 minutes for color to be produced.
- 4. Record results.

#### **Safety Concerns:**

Always wear eye protection, gloves and a laboratory coat when preparing this reagent.

Eye protection and a laboratory coat should be worn when using this reagent for color tests.

Sulfuric acid is a strong oxidizing agent and corrosive.

# **Literature References:**

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Butler, William P., Methods of Analysis, IRS Publication #341, 1966, p. 136.

Johns, S.H., "Spot Tests: A Color Chart Reference for Forensic Chemists", **Journal of Forensic Science**, July, 1979, pp. 631-649.

This procedure has been used in the Drug Chemistry Section since 1971.