



**TRAINING PROGRAM FOR BODY FLUID IDENTIFICATION**  
**NORTH CAROLINA STATE BUREAU OF INVESTIGATION**

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## **1. PURPOSE AND SCOPE**

The purpose of this manual is to provide a consistent training program for the analysis of forensic evidence by the Molecular Genetics Section at the North Carolina State Bureau of Investigation. This program provides individuals with the theoretical background and the working knowledge to conduct independent casework analysis and effective expert witness testimony in the field of body fluid identification. Heavy emphasis shall be placed on quality assurance of all tests performed, data integrity via thorough documentation, and excellence in obtaining consistent and congruous results.

The training program detailed in this document provides the following:

- Theoretical knowledge of the principles of body fluid identification testing.
- Working knowledge of the principles and practices of serological theories such as antigen antibody reactions, immunodiffusion, etc. as they relate to the forensic analysis of body fluids.
- The ability to perform independent, accurate, and consistent forensic analysis on forensic case material.
- The ability to provide effective expert witness testimony that includes, but is not limited to the presentation of body fluid identification technology and the defense of analytical conclusions.

## **2. REQUIREMENTS FOR QUALIFICATION**

### **2.1 Prerequisites**

Individuals must possess a strong scientific background and have course work in biology, chemistry, biochemistry and genetics. Individuals must complete the following courses at NCSU or their



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equivalent: Genetics 501, 502 (now combined to form Genetics 513), and either Genetics 560 or Biochemistry/Genetics 561.

## 2.2 Competency Tests

Individuals must pass a series of well defined competency tests. These tests are to determine the trainee's ability to consistently analyze body fluids from a variety of sources. Proficiency tests will include blood, mixed fluids, and simulated cases.

## 2.3 Written Examination

A written examination shall examine the trainee's understanding of the theoretical and working knowledge of body fluid identification tests and how they are applied in the laboratory.

## 2.4 Mock Trial

A mock trial shall be used to determine the trainee's ability to provide effective expert witness testimony.

# 3. **INSTRUCTIONS FOR THE TRAINING OFFICER**

3.1 This program is designed to provide each trainee with the theoretical background and working knowledge to reliably analyze forensic material utilizing body fluid identification tests. Every topic listed in this manual is equally important; a deficiency in one area can lead to the failure of a successful analysis and/or defense of the analysis in a court of law. Therefore, the training officer must pay very close attention to detail and ensure that all quality assurance guidelines are being followed for every sample processed in the training program. By ensuring each trainee maintains a high degree of concentration and awareness during the performance of his/her training, the proper techniques will be learned and later successfully applied to actual casework.

3.2 The order of topics listed in this manual are not necessarily in the chronological order that the tests will be performed. It may be necessary to learn



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and perform some techniques out of order. In this case, it is the responsibility of the training officer to provide the trainee with a clear explanation of any missing points or steps and later logically tie everything together.

3.3 It is the responsibility of the Training Officer to point out pertinent scientific literature and technical manuals included in the bibliography to the trainee so that they may become familiar with these readings.

3.4 It is the responsibility of the Training Officer to explain potential safety hazards to each Trainee BEFORE performing a task that may involve said potential safety hazard.

#### **4. INSTRUCTIONS FOR THE TRAINEE**

4.1 The trainee is required to keep files on all work completed. These files should include but are not limited to the Training Manual Log Sheet, worksheets, and copies of reports issued in cases they participate in. These files will be checked periodically by the Training Officer and/or SAC.

4.2 The readings assigned are very important. While it is not necessary to memorize reagent recipes, it is necessary to become familiar with the principles of each test and the protocol and be able to perform all duties independently. The trainee is expected to become familiar with the literature that pertains to the forensic analysis of body fluid identification testing that is included in the bibliography.

#### **5. SAFETY ISSUES, INTRODUCTION TO THE LAB AND ORGANIZATION**

5.1 There are many potential hazards that exist in the laboratory. While the exposure to all hazards can be minimized or avoided, it is the responsibility of the Training Officer to ensure the Trainee is aware of all potential hazards. These potential hazards include but are not limited to the following:

##### Infectious Agents

- A. Viral agents, including HIV and Hepatitis
- B. Bacteria, including sexually transmitted diseases
- C. Fungi
- D. Parasites

##### Hazardous Materials

- A. Caustic Agents (Acids and Bases)



- B. Carcinogens/Mutagens
- C. Teratogens
- D. Organic Chemicals

Electrical Hazards

Shock from any piece of electrical equipment

Burn Hazards

Autoclaves      Bunsen burners  
Sterilizers      Hot Plates

Eye Damage

Alternate light sources

5.2 Laboratory Safety Procedures

5.2.1 Individuals must be trained in laboratory safety by the Section Safety Officer prior to the commencement of training. Various manuals are provided that must be followed to ensure safety of all laboratory personnel. The following manuals are to be used for reference and guidance for laboratory safety: MSDS Notebook, Molecular Genetics Section Manuals, Chemical Hygiene Program, Bloodborne Pathogen Program, and the DOJ Safety Manual. The trainee will also be briefed on the fire evacuation plan for the laboratory.

5.2.2 It is the responsibility of the training officer to alert the trainee to safety hazards specific to this laboratory, including all items mentioned in 5.1.

5.3 Laboratory Orientation

5.3.1 The trainee will be taken throughout the laboratory and shown areas of interest to their work. They will also be provided with a written job description, an organizational chart, and various manuals including the Crime Laboratory Procedures Manual, SBI Policy Manual, and Directives for both manuals, DAB Standards, TWGDAM QA Guidelines, and the Section Quality Assurance Manual.

**6. ASEPTIC TECHNIQUE AND CONTAMINATION CONTROL**



6.1 This Section uses the Polymerase Chain Reaction (PCR) technology which allows very small amounts of DNA to be amplified over a million times. Because of the sensitivity of this technique, contamination control is a very serious issue that must be emphasized and practiced with every sample, starting with the identification of the body fluids present on evidence. The Evidence Handling Procedures must be strictly followed.

6.1.1 All items used in the identification, transfer and isolation of forensic samples must be sterile and/or free of contaminate DNA.

6.1.2 Gloves must be worn at all times while handling samples. This is to protect both the analyst and sample.

6.1.3 A fresh, sterile pipette tip must be used for each transfer of fluid or chemical to be used for DNA analysis.

6.1.4 Special precautions must be taken to ensure that the bench and surrounding areas have been properly decontaminated.

6.1.5 Scissors, tweezers, and other instruments used for cuttings or extractions shall be sterilized in between each sample.

6.1.6 The preparation of the known and unknown samples shall be separated by time.

## **7. DOCUMENTATION**

### **7.1 Goals**

7.1 To provide protocols for the preparation of reagents and performance of tests to ensure consistent, reliable results.

7.2 To provide a thorough record of events for each case analysis.

### **7.2 Protocol Notebook**



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The Body Fluid Identification Procedures Manual shall be made available to each analyst. The analyst shall not deviate from any procedure without permission from the SAC. Any deviation (purposely or by mistake) from the protocol shall be thoroughly documented on the worksheet at the time of occurrence.

7.3 Buffers and Solutions Recipe Book

The working copy of the appropriate QC forms, which includes procedures for preparing solutions, will be maintained in the QC notebook.

7.4 Worksheets

The purpose of the worksheets are to provide a means to thoroughly document each step of the analytical process. Each worksheet is to be completed either during or as soon as possible following the step.

**8. RECEIVING AND HANDLING OF EVIDENCE**

8.1 Goals

- 8.1.1 To obtain a working knowledge of factors and conditions that influence the deterioration of evidence as it relates to packaging, handling, and storage conditions and time.
- 8.1.2 To develop a thorough understanding of evidence handling procedures.
- 8.1.3 To develop a thorough understanding of the necessity for detailed comprehensive notes and adequate labeling of evidential materials.

8.2 Tasks

- 8.2.1 Read and become thoroughly familiar with the SBI Evidence Handling Procedures regarding receiving, identifying, and handling of evidence; as well as specific guidelines for handling biological evidence within the Section.

**9. ANALYSIS OF BODY FLUID SAMPLES**

9.1 Goals



- 9.1.1 To develop a basic understanding of the methodology and theory of chemical, microscopic, and immunological testing procedures used to identify blood and the species from which the blood may have originated.
  - 9.1.2 To develop a basic understanding of the methodology and theory of chemical, microscopic, and immunological testing procedures used to identify semen.
  - 9.1.3 To develop a basic understanding of the methodology and theory of chemical testing procedures used to identify amylase.
  - 9.1.4 To develop skills that will allow the trainee to independently and successfully analyze forensic samples.
  - 9.1.5 To become familiar with the sensitivity and limitations of the procedures used.
  - 9.1.6 To develop a cognizant understanding of contamination issues and the steps necessary to avoid contamination.
  - 9.1.7 To understand the use of controls during each procedure.
  - 9.1.8 To become familiar with and understand the function of any buffers, solutions, or reagents used.
  - 9.1.9 To become familiar with all documentation required.
- 9.2 Tasks
- 9.2.1 Prepare any buffers, solutions, or reagents necessary for testing.
  - 9.2.1 Understand and perform quality control checks necessary on buffers, solutions, reagents and test kits used.
  - 9.2.3 Perform testing on various sample types which allow for ample testing for each analytical procedure used. The training samples will represent materials commonly encountered in forensic casework. The Training Officer will initially present the trainee with a short lecture on the analytical procedure in question. The trainee is then given samples to practice on. Once a block of instruction is completed, the trainee will receive a written test and a set of





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competency samples to analyze. The exact number of samples will be determined by the training officer in accordance with experience and ability of the trainee. Upon successful completion of the block of instruction, the trainee will start on the next block of instruction.

9.2.4 Blocks of instruction the trainee will complete to identify blood include:

9.2.4.1 Phenolphthalein testing (presumptive chemical tests)  
Luminol testing - lecture and crime scene experience

9.2.4.2 Takayama (microcrystalline identification testing)

9.2.4.3 Species origin testing (immunological methods)

9.2.5 Blocks of instruction the trainee will complete to identify semen include:

9.2.5.1 Acid Phosphatase testing (presumptive chemical tests)

9.2.5.2 Spermatozoa identification (microscopic identification)

9.2.5.3 P30 testing (immunological and/or electrophoretic methods)

9.2.6 Blocks of instruction the trainee will complete to identify amylase include:

9.2.6.1 Phedebas testing (presumptive chemical tests)

9.2.7 Complete documentation for all procedures.

## **10. REPORT WRITING AND TESTIMONY**

### 10.1 Goals

10.1.1 To develop the skills necessary to effectively report body fluid identification results, and provide expert witness testimony in a court of law.

10.1.2 To develop a working knowledge of the terminology and presentation of analysis and results.



10.1.3 To become skilled in expressing written and oral results simply, concisely, and accurately.

10.1.4 To become familiar with pertinent scientific literature.

10.2 Tasks

10.2.1 Read and understand pertinent scientific literature provided in the Bibliography (Appendix III).

10.2.2 Thoroughly understand and be able to accurately and concisely answer the questions in Appendix II.

10.2.3 Draft reports on analysis of proficiency tests results.

10.2.4 Draft reports from cases previously analyzed by the section.

10.2.5 Attendance at court to observe qualified examiners testify.

10.2.6 Successful completion of a moot court.

11. **COMPETENCY TESTING**

11.1. Goals

11.1.1 Upon successful completion of all blocks of instruction, the trainee will be given a series of competency tests that closely mimic forensic evidence. The trainee must score 100% accuracy in these tests. This test is the final one that the trainee must complete before being allowed to work on forensic cases.

11.2 Tasks

11.2.1 Successful completion of the competency tests.

12. **CASE INTERNSHIP PROGRAM**

12.1 Goals

12.1.1 Once the trainee completes their competency tests, they are placed in a case internship program with qualified examiners. The purpose of this



internship is to allow the newly qualified examiner to work cases, but under the close supervision of a qualified examiner. The goal of this program is to allow the newly qualified examiner to work their first cases under the direct supervision of a qualified examiner and gain confidence in their analytical skills.

## 12.2 Tasks

- 12.2.1 The newly qualified examiner will conduct the entire analysis of a more experienced examiner's case, prepare the notes, and will write the report, just as if the case was theirs to work.
- 12.2.2 The more experienced examiner will observe all results, initial the notes, and evidence, and review all work conducted. They will constructively critique the work conducted by the newly qualified examiner and ask them questions about the case that they could expect in court. The case will be assigned to the experienced examiner, and the experienced examiner will sign the laboratory report. The laboratory notes cover sheet will contain a note that the case was worked by \_\_\_\_\_ as part of their case internship program.
- 12.2.3 The case internship program will last no less than one month. The newly qualified examiner, the experienced examiners, and the Special Agent In Charge will all decide when the analyst is ready for unsupervised casework analysis.



**APPENDIX I. NORTH CAROLINA STATE BUREAU OF INVESTIGATION:  
Body Fluid Identification Training Manual Log Sheet**

<b>Training Area</b>	<b>Date Completed/Initials</b>	<b>Trainers Initials</b>
<b>Education</b>		
Safety		
1. Chemical Hazards/MSDS Sheets		
2. Electrical Safety		
3. Blood borne Pathogen Training		
4. Readings		
Laboratory Orientation and Organization		
1. Lecture		
2. Readings		
Blood Identification		
1. Lecture		
2. Readings		
Semen Identification		
1. Lecture		
2. Readings		
Saliva Identification		
1. Lecture		

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2. Readings		
Quality Control and Documentation		
1. Use of QC Reagent Book		
<b>Training Area (continued)</b>	<b>Date Completed/Initials</b>	<b>Trainers Initials</b>
2. Use of QC forms		
3. Documentation using Worksheets		
Receiving and Handling Evidence		
1. Read SBI Evidence Procedures		
<b>Laboratory Training</b>		
Aseptic Technique and Contamination Control		
1. Cleaning of Equipment		
2. Use of Biosafety Hood		
3. Handling of Samples		
Blood Identification		
1. Presumptive Testing		
1.1 Read and Understand Procedure		
1.2 Demonstration of Test		
1.3 Supervised Testing of Known Samples		
1.4 Competency testing		
2. Confirmatory Testing		
2.1 Read and Understand Procedure		
2.2 Demonstration of Test		
2.3 Supervised Testing of Known Samples		



2.4 Competency Testing		
Species Origin Testing		
<b>Training Area (continued)</b>	<b>Date Completed/Initials</b>	<b>Trainers Initials</b>
3.1. Read and Understand Procedures		
3.2. Demonstration of Tests		
3.3 Supervised Testing of Known Samples		
3.4 Competency Testing		
Semen Identification		
1. Presumptive Testing		
1.1 Read and Understand Procedures		
1.2 Demonstration of Tests		
1.3 Supervised Testing of Known Samples		
1.4 Competency Testing		
2. Confirmatory Testing		
2.1 Read and Understand Procedures		
2.2 Demonstration of Tests		
2.3 Supervised Testing of Known Samples		
2.4 Competency Testing		
Saliva Identification		
1. Presumptive Testing		
1.1 Read and Understand Procedure		
1.2 Demonstration of Test		
1.3 Supervised Testing of Known Samples		

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1.4 Competency Testing		
<b>Training Area (continued)</b>	<b>Date Completed/Initials</b>	<b>Trainers Initials</b>
<b>Report Writing and Testimony</b>		
able to Draft a Satisfactory Report from Previously Analyzed Cases		
Witness Expert Testimony		
Familiar with Court Procedures		
Successful Completion of a Moot Court		
<b>Competency Testing</b>		
Successful Completion of Practical Competency Tests		
Successful Completion of Final Written Test		
<b>Case Internship Program</b>		
Successful Completion of a Case Internship Program		



**APPENDIX II. NORTH CAROLINA STATE BUREAU OF INVESTIGATION:**

1. Describe the basis for presumptive tests for blood in general.
2. Name two classes of substances that react with the presumptive tests other than blood.
3. Why is a three step procedure better than a one step blood testing procedure?
4. When, why, and how is luminol used instead of the other chemical tests?
5. What substances will cause a false positive reaction with luminol?
6. Name the components of the Takayama test and explain the role of each.
7. Will commercially prepared catalase and peroxidase will give a positive result with the Takayama reagent?
8. Describe how the Takayama test is performed on a bloodstain.
9. Describe the appearance of Takayama crystals.
10. What is the biochemical basis for species identification by Ouchterlony testing?
11. How is anti-serum made?
12. List the advantages and disadvantages to Ouchterlony testing.
13. List and explain the appropriate controls used in Ouchterlony species identification.
14. What environmental factors affect the stability of a bloodstain?
15. How does heat, such as autoclaving, affect a bloodstain?
16. Which is more sensitive, a blood presumptive test or a human origin test?
17. What is semen and list as many components that you can think of that are found in semen? Specify which are semen specific and which are not.
18. List three sources of Acid Phosphatase.
19. What components identify semen conclusively?
20. Briefly describe and draw a spermatozoa labeling the major parts.
21. Name the staining procedure used for sperm identification and identify the names of the stains that we use.
22. Are sperm haploid or diploid?
23. What does azoospermic mean? How does a person become azoospermic (list three)?
24. What is p30?
25. How do we detect p30? Describe the controls used in each test.
26. Describe the principle of the Phadebas testing procedure.
27. Describe the procedure for testing saliva with Phadebas tablets.
28. Is Phadebas testing a confirmatory or presumptive test for saliva?

**APPENDIX III. NORTH CAROLINA STATE BUREAU OF INVESTIGATION:  
Background and Reference Information**





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## 1. Bloodstain Identification

### 1.1 Presumptive tests

Presumptive tests, or catalytic tests, for blood center on the erythrocyte portion of the formed elements. Each of these red blood cells carries approximately 280 million molecules of hemoglobin, each possessing four heme units. A heme compound is represented by an iron center coordinated with four nitrogen containing compounds, each of which bind to one oxygen molecule to transport it in the circulatory system. This heme group acts like a peroxidase, an enzyme which can catalyze the oxidation of an organic compound by utilizing peroxide. While heme is actually a conjugated protein, and not an enzyme, it acts like a peroxidase to dissociate hydrogen peroxide into two hydroxyl free radicals, which are highly reactive and tend to oxidize organic substances. In catalytic testing, these organic compounds are color reagents which change in color when transforming from the reduced to the oxidized state. This technique allows for a quick visual screening of blood but should not be judged as a confirmation of the presence of blood. Presumptive tests are designed to be used in conjunction with confirmatory tests for blood if enough sample is available.

#### 1.1.1 Luminol

##### Background

In 1928 Albrecht reported on the chemiluminescent properties of the compound 3-aminophthalhydrazide. In 1935 Hundress, Stanley, and Parker published a method for the [reparation of the compound and named it luminol. In 1937 Specht published his findings on the application of lumionol to the forensic field. It was through his studies that the use of luminol as a chemiluminescent test for the presence of blood was advocated.

Luminol is a chemiluminescent presumptive test for the presence of blood. Although the exact mechanism of conversion from the reduced to oxidized state is not well understood, it is generally recognized that two tautomeric forms of the oxidized structure create the chemiluminescent qualities.

Once sprayed on a surface, a bluish light emission may be detected at concentrations as low as 0.1 PPM. Luminol is employed when no visible blood is detected or other less sensitive presumptive tests have failed. It is also primarily used for large items such as cars and houses.



The analyst should be particularly aware that false positives will occur on vegetable peroxidases, some metals and chemicals. Therefore, care should be taken when interpreting the test results. Patterns visualized during the use of luminol should be photographed if possible.

#### Luminol Photography:

Luminol reactions must be photographed in total darkness so that the only light to strike the film comes from the luminol reaction. To photograph the reactions follow the steps below:

1. Load a Nikon 35 mm camera with Tri-X Pan film and set the film speed (ASA) at 400.
2. Use a 50 mm f2 lens and set the aperture to its widest opening. Adjust the shutter speed to the "B" setting.
3. Mount the camera on a tripod in such a way that the lens is facing the object being photographed. Adjust the tripod length so that the entire viewfinder frame is used for the object photographed.
4. Place a six inch ruler with fluorescent tape at one inch intervals next to the pattern being photographed so that a 1:1 photograph can be made for direct comparison to a known object.
5. Use a shutter release cable. To maintain enough light, and thus a visual pattern, lightly spray the area with luminol while the shutter is open. Avoid over-spraying the area, spray just enough to see the pattern.
6. To ensure a good quality photograph, use varying exposure times. These exposure times will vary depending on how strong the reaction is.

#### References

Gaensslen, Ren (1989) Sourcebook in Forensic Serology, Immunology, and Biochemistry, 2nd ed., National Institutes of Justice, pp. 112-114



Chang, CA, Patterson HH (1980) Halide ion enhancement of chromium (III), Iron (II) and cobalt (II) catalysis of luminol chemiluminescence. *Analytical Chemistry* 52: 653-656

Lee, H (1982) Identification and Grouping of Bloodstains in R. Saferstein, ed., Forensic Science Handbook, Vol. 1, Prentice Hall, Englewood Cliffs, N.J., pp. 283-297

Lytle LT, Hedgecock DG. (1978) Chemiluminescence in the visualization of forensic bloodstains, *Journal of Forensic Sciences*, 23: 550-562

Hunt AC, Corby C, Dodd BE, Camps FE (1960) The identification of human bloodstains. *Journal of Forensic Medicine* 7:112-130

Proescher F, Moody AM (1939) Detection of blood by means of chemiluminescence. *The Journal of Laboratory and Clinical Medicine* pp 1183-1189

White EH, Roswell DF (1970) The chemiluminescence of organic hydrazides. *Journal of the American Chemical Society* 3:54-62

Zweidinger RA, Lytle LT, Pitt cG (1973) Photography of bloodstains visualized by luminol. *Journal of Forensic Science* 18: 296-302

#### Training Tasks:

1. Completion of the section and Bureau crime scene training course.
2. Assist in the processing of approximately 20 crime scenes under the supervision of a trained examiner.

### 1.1.2 Phenolphthalein test

#### Background

The phenolphthalein test is a presumptive catalytic test for the presence of blood. The heme portion of hemoglobin possess a peroxidase-like activity which catalyzes the breakdown of hydrogen peroxide into free hydroxyl



radicals. These hydrozyl radicals then oxidize the reduced phenolphthalin, producing a pink color.

To maintain this pink color, the reaction is carried out in alkaline conditions. If the pH is lowered to acidic conditions, the solution will again become colorless, but is in the form of phenolphthalein. The working solution is maintained in zinc to keep phenolphthalin in the reduced state. Use care in the preparation of the phenolphthalin, since flammable gas may be generated. For this reason an electric heating mantle is used. Also be cautious since zinc dust, in the presence of water, may act as a fire catalyst.

This test is particularly useful because there are less known false positives than other presumptive tests. The literature reports that certain plants including horseradish, tomato, turnip, and Jerusalem artichoke possess elevated levels of peroxidase which may give a positive reaction with phenolphthalin. This false reaction may be eliminated by heating the filter paper to 100°C for 30 minutes, which destroys the peroxidase activity. The literature also reports that bacteria which possess a high catalase activity may give a false positive reaction. If a pink color appears after the addition of phenolphthalin to the filter paper, but before the addition of the hydrogen peroxide, then the presence of an oxidant is indicated. Any reaction that occurs more than 5 seconds after the addition of the hydrogen peroxide is considered a false positive and is not recorded. Metals and rust do not interfere with this testing. However, it may be slightly less sensitive than some other catalytic tests.

#### References

Gaensslen, RE ( 1989) Sourcebook in Forensic Serology, Immunology, and Biochemistry, 2nd ed., National Institutes of Justice, pp 103-105

Lee, HC ( 1982) Identification and Grouping of Bloodstains, in R. Saferstein, ed., Forensic Science Handbook, Vol. 1, Prentice Hall, Englewood Cliffs, N.J., pp 283-297

Blake, ET, Dillon DJ (1973) Microorganisms and the presumptive tests for blood. *Journal of Police Science and Administration* 1: 395-400

Higaki RS, Philp WMS (1976) A study of the sensitivity, stability, and specificity of phenolphthalein as an indicator test for blood. *Canadian Society of Forensic Sciences* 9: 97-102



Training Tasks:

1. Testing the effects of washing and heating on blood detection

Use fresh blood and make a stain on clean cotton sheeting. Subject a portion of the stain to each of the following conditions:

- a. Wash in a washing machine and dry in the dryer
- b. Wash in cold water in a sink, air dry
- c. Wash in hot water in the sink, air dry
- d. Wash by hand with soap, air dry
- e. Soak in water overnight, air dry
- f. Dry in a dryer, but don not wash first
- g. Char on a microscope slide
- h. Heat to 100° C
- i. Iron the stain

2. Sensitivity Testing

Take a fresh liquid bloodstain and make serial dilutions. Make stains from the dilutions on clean cotton sheeting and then air dry them. Test each stain with phenolphthalein.

3. Specificity Testing

Collect the following vegetables and rub them thickly over filter papers and test with phenolphthalein.

Horseradish	Pineapple	Red Grapes	Red Cabbage	Cantaloupe
Radish	Celery	Onions	Spinach	Lettuce
Carrots	Broccoli	Tomatoes	Peas	Raisins
Mushrooms	Artichoke	Turnip	Ketchup	Cucumber

Also collect and test cola stains, chicken blood, beef blood (from meats).

## 1.2. Takayama test (Confirmatory Testing)

### Background

The Takayama test will confirm the presence of blood and is designed to be used in conjunction with presumptive testing for blood. Hemoglobin is composed of a heme prostetic group and globin. When a blood stain dries, the iron in the heme group is in the ferric state (FE+3) due to the formation



of methemoglobin. In order for the Takayama test to work, the dried stain must be hydrolyzed to metheme and globin via alkaline hydrolysis by sodium hydroxide. Next the iron in the ferric (Fe +3) state must be reduced to the ferrous state (Fe +2) via the use of glucose, a reducing sugar. Once in the ferrous state, the iron will combine with pyridine to form pyridineferroprotoporphyrin, which is an insoluble crystalline product. A positive result is visualized microscopically by the formation of salmon colored rhomboidal or stellate crystals

The only materials that will give a positive reaction other than blood are commercially produced preparations of catalase and peroxidase, items not occurring in nature.

References:

Gaensslen, RE, (1989) Sourcebook in Forensic Serology, Immunology, and Biochemistry, 2nd ed., National Institutes of Justice, pp. 85-87.

Lee, HC, (1982) Identification and Grouping of Bloodstains, in R. Saferstein, ed., Forensic Science Handbook, Vol. 1, Prentice Hall, Englewood Cliffs, N.J., pp. 283-297

Training Tasks:

1. Test the materials prepared for phenolphthalein testing and record the results.
2. Test purified catalase and peroxidase and old bloodstains provided by the training officer.
3. Place blood on various matrices, absorb it off and test it.

**1.3. Human (or Species) Origin test - via Ring Precipitin, Ouchterlony, or Anti-human Hemoglobin Test Methods**

Background

The precipitin test is one method of distinguishing between human and animal blood. It utilizes the biological properties of antibody-antigen complex formation to allow a visual representation of a reaction. Antibodies are very large molecules and are represented by five classes of immunoglobulins, IgG, IgA, IgM, IgD, and IgE. An antigen is a substance which has the ability to produce an immunological response when introduced into a foreign animal. The antibodies are produced by a



host animal when the animal is injected with a foreign protein containing antigens and the host becomes sensitized. The immune system of the host recognizes the foreign antigen and produces antibodies to react with it in a very specific manner.

In common forensic testing, the antibodies to human antigens are raised in rabbits which results in rabbit anti-human antiserum. Dr. Uhlenhuth in 1901, presented evidence of the specificity for human antigens to only agglutinate with complementary antibodies. Therefore, if the antibodies in the anti-human antiserum comes in contact with human antigens, the specificity of the reaction allows for the formation of the human antigen-antibody complex and prohibits the formation of other non-specific complexes. In addition to testing for human antigens, the same test can be performed on a variety of animals. For example, goat anti-swine antiserum can be used to determine if a blood sample originated from a pig.

It should be noted that there is some anti-sera cross reactivity between the antigens in a closely related species. For example, ant-ram sera may cross react with goat and cow antigens. In humans, some monkeys or higher primates may produce a response. The Cappel rabbit antiserum to human serum will produce (i) a positive reaction in a 1:3 dilution of monkey blood in a 24 hour test (ii) a positive reaction up to a 1:100 dilution for human blood in a 24 hour test and (iii) a positive reaction up to 1:1000 for human blood in a 48 hour test.

The antigen-antibody reaction occurs in two steps. The first is called sensitization where the antigen and antibody form weak bonds, probably a combination of electrostatic, hydrophobic, and vander Waals interactions, and coordinate themselves for phase two. In this step, the complex is formed and the lattice structure begins to multiply which is representative of precipitation. This visualization step creates a white band which can be recognized for analysis.

Several methods have been developed to monitor the formation of the antigen-antibody complex including the Ring Test, Single Diffusion in One or Two Dimensions, or Double Diffusion in One or Two Dimensions. The latter is also referred to as the Ouchterlony, or immunodiffusion, method. While the different tests exhibit a range of sensitivity, some general characteristics are the same due to the antigen-antibody interaction mechanism.



A blood sample can fail to produce a precipitin band, also known as a false negative. This may occur if the sample is degraded due to age, heat, sunlight, chemical treatment with detergents, aluminum oxide, pulverized iron ore, or when mixed with some types of soil. Also note that the complex is best formed when the antigen and antibody are present in approximately equal concentrations. Either component present in excess can result in a weak reaction or a false negative.

With certain tests, several substances have been documented to exhibit a reaction which may mimic a true antigen-antibody reaction, also known as a false positive. These include aluminum and iron chlorates, aluminum chromate, salts of alkyl sulfonates and alkyl sulfates, peroxide, some dilute acids and bases, tannic acid, and spruce bark extract.

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#### Training Tasks:

1. Test a wide variety of human and animal bloods that will be provided by the training officer.

## 2. Seminal Stain Identification





Semen testing is comprised of essentially two different parts. These are a presumptive test for the presence of acid phosphatase, which is used as a preliminary screening and confirmatory testing, including sperm searches and detection of p30, the latter of which is used primarily when no sperm are detected.

Semen is a white gelatinous liquid ejaculated by the male. It consists of water, flavins, citric acid, fructose, other sugars, spermine, free amino acids, phosphorycholine, acid phosphatase, prostoglandins, and sperm (except in vasectomized and aspermic males). There are four portions of the ejaculate. The first is the alkaline secretions of the Cowpers (Bulbourethral) gland, which clear the urethra and neutralize the acidic environment. The second portion originates from the prostate gland and contains alkaline secretions that are rich in proteins and proteolytic enzymes, and acid phosphatase and p30. The third portion is the sperm cell fraction and the final is the seminal plasma which is produced in the seminal vesicles.

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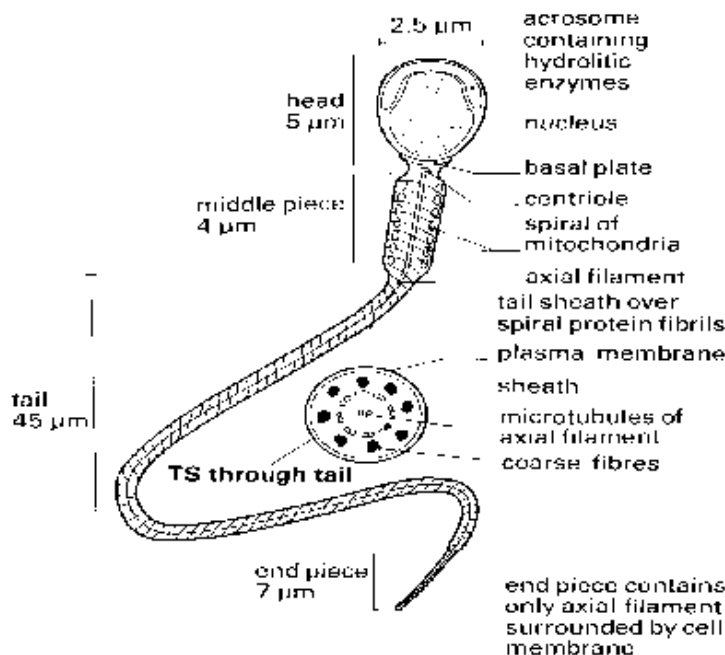
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## 2.1 Acid Phosphatase Presumptive Chemical Testing

### Background

The detection of acid phosphatase (AP) for use as a presumptive test for the presence of semen is based on the observation that levels of AP were 200 times higher in semen stains than in any other material tested. Seminal acid phosphatase (SAP) is formed in males and is a product of the prostate gland, whose arrangement in the reproductive system, allows for the possibility of the presence of SAP without spermatozoa. Conditions including vasectomy, and exposure to drugs and chemicals can reduce sperm counts. Production of SAP is tied to testosterone, which means that lower levels of SAP are probable in adolescent males. It has also been noted that increased levels are present in prostatic cancer patients.

In general, AP concentration is higher in the area surrounding a stain and spermatozoa are concentrated more in the central region. The theory and reaction used in AP testing is based on the biochemical aspect of AP as an enzyme. When presented with sodium alpha naphthyl phosphate, AP will cleave the sodium phosphate group and produce the alcohol of the fused ring structure, naphthol. Naphthol will then react with Brentamine fast blue B and the azo group at the carbon adjacent to the hydroxyl to create a purple color. There are several other substrates which react via the same mechanism of phosphate removal and formation of an azo dye, including phenylphosphate, p-nitro phosphate, and thymolphthalein monophosphate.



False positives may occur on substances such as human milk, some vegetable extracts, and various other substances rarely encountered in forensic casework. Some vaginal treatments such as deodorants or spermicides could lead to false negatives.

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**2.2 Christmas Tree Stain (PICS and NFR)**

Background

Sperm cells can be stained by several techniques including carmine, iodine-KI mixture, and methyl blue. However, the most common method is referred to as the Christmas tree technique and utilizes nuclear fast red (NFR) and picroindigocarmine (PCI) as stains. In this procedure, the sperm heads are stained red due to the intercalation of nuclear fast red into the nucleus of the sperm head. A clear acrosomal cap is also noted. The picroindigocarmine is used to stain the cytoplasm of the epithelial cells green to increase the contrast between the two most predominant cell types in forensic casework.



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## 2.3 P30 Analysis



Sensabaugh identified, in 1973, a 30,000 MW antigenic protein in seminal fluid which he labeled p30. It was discovered when a reaction occurred between an antiserum prepared by Li and Beling. p30, also known as Prostate Specific Antigen (PSA), is present in the epithelial cells of the prostate ducts, so it can be present regardless of the presence of spermatozoa. Since it is a product of the prostate, there is none present in females. Levels of p30 concentration vary, but in a study by Brown, in one ml of semen, levels ranged from 300 to 4200 micrograms, with a mean of 1200 micrograms. In one ml of blood, the mean concentration of p30 is 0.0026 micrograms, unless the patient is inflicted with prostatic cancer where levels are in the 0.5 microgram range. This demonstrates that the concentration is significantly higher in semen than blood or other body fluids. Baechtel reported that prostatic tissue, prostatic carcinoma, and urine may also contain some level of p30 for obvious reasons. PSA has also been detected in orangutan and macaque semen, but these samples also had much lower levels of SAP, and could be distinguished from human serum by AP testing.

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Proceedings of the International Symposium on Forensic Immunology, US Department of Justice, 1986

#### Abacard

##### Background

This test detects an antigen-antibody reaction. A small extract of a stain is placed into a well and the sample migrates across a strip to the test and control areas. If p30 is present in the semen specimen, it will react with the mobile monoclonal antihuman p30 antibody and a complex is formed. A polyclonal antihuman p30 antibody is immobilized on the membrane in the



test area which captures the complex so that an antibody-antigen-antibody sandwich is formed. The dye particle-p30 complex concentrate in the test area. When the p30 concentration in the sample exceeds 4 ng/ml the pink dye particles will form a pink colored band. The p30 antibody-dye conjugates cannot bind to the antibody in the test area, but are captured by an immobilized anti immunoglobulin antibody present in the control area. A pink band in this area indicates that the test performed properly. Therefore, a pink line must be formed in both the test and control areas for a positive result.

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## 2.4 Chemical Indications of Saliva

Amylase, namely alpha amylase for forensic purposes, is an enzyme found in plants and animals which hydrolyzes glucose polymers at the C<sup>1</sup>-O<sup>4</sup> bond of



the alpha 1,4 linkage. The amylase found in human saliva is of this alpha form, and this hydrolyzing characteristic is used to detect for the presence of saliva stains. Forensic methods generally use starch as the substrate. Starch is made up from amylose and amylopectin, both of which contain the alpha 1,4 linkage which can be hydrolyzed by salivary amylase. However, amylopectin contains the glucose side chain bound by an alpha 1,6 linkage which cannot be broken. Therefore, the amylopectin is not degraded completely.

A test for amylase is only presumptive for the presence of saliva because it is found in other body fluids. However, the concentration of amylase is more than 600 times more concentrated on average in saliva than sweat, the next highest amylase containing component of the bodily fluids.

### Phadebas test

#### Background

This method of detecting amylase utilizes a water –insoluble cross-linked starch polymer carrying a blue dye as a substrate. It is hydrolyzed by alpha amylase to form water soluble blue fragments. The amount of amylase present in the sample is proportional to the concentration of blue dye liberated during analysis. The concentration of blue dye in the supernatant can be analyzed for absorbance with a spectrophotometer or it can be used only qualitatively. For forensic purposes, it is generally sufficient to use the Phadebas tablets to test for positive or negative results.

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